THE AMOUNT OF LICHEN SECONDARY METABOLITES IN *CLADONIA FOLIACEA* (CLADONIACEAE, LICHENISED ASCOMYCOTA)

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The terricolous species *Cladonia foliacea* (Cladoniaceae, lichenised Ascomycota) widely distributed in open, dry lowland steppe and rocky mountain grassland vegetation in Europe was chosen as a potential test organism for ecological experiments, since their thalli are producing cortical solar radiation-protective and UV screening pigment dibenzofuran usnic acid and medullary secondary substance depsidone fumarprotocetraric acid. Significant seasonal differences were found in the amounts of lichen secondary metabolites analysed by HPTLC and HPLC-PDA between summer and winter collected thalli in sandy grassland area in Hungary. The concentrations of usnic acid varied between 7.34 and 15.52 mg/g in summer collected samples and 13.90 and 21.61 mg/g in winter collected ones. A comparable amount (11.61±0.29 mg/g) was measured in pulverised samples. The concentrations of fumarprotocetraric acid varied between 0.60 and 3.01 mg/g in summer collected samples and 2.26 and 5.81 mg/g in winter collected thalli. A comparable amount (2.45±0.21 mg/g) was found in pulverised samples. The range of concentration values is comparable with data known from lichens. A higher amount of usnic acid is produced in winter probably to ensure sufficient protection also for summer. The fumarprotocetraric acid content of the medulla might contribute to the solar irradiation reflecting role of the pale lower surface lobes turning upwards in dry condition.

Key words: acetone rinsing, chlorolichens, fumarprotocetraric acid, high performance liquid chromatography with photodiode array detector (HPLC-PDA), seasonal differences, usnic acid

INTRODUCTION

Lichenised associations represent a sensitive model ecosystem for investigating various or changed environmental conditions (cf. Asplund *et al.* 2017, Bergamini *et al.* 2005, Galloway 1993, Nash 2008, Stofer *et al.* 2006). Lichens produce a great variety of lichen secondary metabolites (LSMs), and most of them are unique to lichen-forming fungi as it is summarised in recent reviews on their biological roles (Calcott *et al.* 2018, Molnár and Farkas 2010).

Approximately 1,050 secondary compounds have been identified to date (Stocker-Wörgötter 2008, 2015). These chemically diverse (aliphatic and aromatic) lichen substances are produced by the mycobiont (Elix 1996, Huneck 1999). Some of them (such as atranorin, parietin, usnic acid or fungal melanins) are accumulated in the cortex, others (such as norstictic acid, physodic acid, physodalic acid, protocetraric acid, etc.) are detected as tiny extracellular crystals within and on the proteinaceous hydrophobic wall surface layer of the hyphae in the medullary and photosynthetic layer (also on photobiont wall surface) (Honegger 1986, 1997, 2012).

Lichen secondary chemistry analyses – and study of their relation to environmental conditions (cf. Armaleo *et al.* 2008, Asplund *et al.* 2017, Hauck 2011, Hauck *et al.* 2009, Matteucci *et al.* 2017, Shukla *et al.* 2016) – clarify the background of the chemical diversity of the studied taxa.

Most of the LSMs can be removed by acetone as it is applied in chromatographic investigations of these substances (Arup *et al.* 1993, Feige *et al.* 1993, Huneck and Yoshimura 1996, Orange *et al.* 2010). According to Solhaug and Gauslaa (1996, 2001) both mycobiont and photobiont were able to survive acetone rinsing treatment. Removing lichen substances from dry lichen thalli by pure acetone – compared to other solvents – is regarded as the least detrimental method (Solhaug and Gauslaa 2001). Acetone rinsed samples continued producing secondary lichen substances (Solhaug and Gauslaa 2004) and pigment composition did not alter after treatment (Candotto Carniel *et al.* 2017).

The lichen species *Cladonia foliacea* is a frequent chlorolichen in open, dry, sun-exposed habitats in the Hungarian lowland steppe areas as well as in low mountain rocky grasslands. Since it has a great potential in experimental applications, it was selected for a test organism and tested for its tolerance for acetone treatment (Farkas *et al.* 2020). Our present aim was to check LSM content of exactly the same thalli by the analysis of the soaking acetone-solution, furthermore to study seasonal differences in samples collected in two seasons, summer and winter.

MATERIALS AND METHODS

Chemicals

All chemicals were of analytical or higher grade. HPLC acetone (VWR) was applied to extract LSMs from intact lichen samples and for chromatographic analysis (HPLC = High Performance Liquid Chromatography, HPTLC = High Performance Thin Layer Chromatography, see also below). Toluene (CARLO ERBA Reagents), acetic acid (Lach-Ner) and sulphuric acid (CARLO ERBA Reagents) were obtained from Reanal for HPTLC investiga-

tions. Acetonitrile (VWR) and ortho-phosphoric acid (Merck) were used for HPLC analysis. Aqueous solutions were prepared with Milli-Q-water (Millipore). Usnic acid was purchased from Sigma Aldrich, fumarprotocetraric acid from Phytolab. Reagents for spot reactions are Pd = p-phenylene-diamine (Sigma Aldrich), K= KOH (Reanal), and C = commercial bleach, for KC reaction K and C applied sequentially, K followed immediately after by C (for preparation of reagents and further details on spot tests see Orange *et al.* 2010).

The research object

Cladonia foliacea (Huds.) Willd. (Cladoniaceae, lichenised Ascomycota) (Fig. 1) is a relatively frequent, terricolous, foliose lichen species in Hungary and widely distributed over Europe both in open, dry and sun-exposed habitats in lowland steppe and mountain grassland communities, also found in northern hemisphere temperate region in North America (Smith *et al.* 2009, Wirth *et al.* 2013). The different LSM content of the cortex (KC+ pale yellow for usnic acid) and the medulla (Pd+ yellow to red for fumarprotocetraric acid) was already indicated by spot tests (Hillmann and Grummann 1957). The thallus has a thickened upper cortex containing usnic acid (Fig. 1) as a protective compound against UV radiation and other effects (e.g. Cochietto *et al.* 2002, Yilmaz *et al.* 2004). The photosynthetic layer contains coccoid green alga photobiont (*Asterochloris* sp., Trebouxiaceae – cf. Moya *et al.* 2015, Peksa and Škaloud 2008, Škaloud and Peksa 2008). The yellowish white, white lower surface lobes are turning upwards when they are dry and reflecting the strong solar irradiation near soil surface. The species also produces a depsidone, the fumarprotocetraric acid (Fig. 1) here in the medullary layer. There is no lower cortical layer developed.

Various effects of these LSMs are investigated in detail analysing extracts of this species (e.g., Anar *et al.* 2013, Aslan *et al.* 2006, Emsen *et al.* 2012, Khadri *et al.* 2019, Koparal 2015, Kosanić *et al.* 2018, Mitrović *et al.* 2011, 2015, Pandir *et al.* 2018, Yilmaz *et al.* 2004).

Collection and treatments of the samples

Lichen thalli were collected in summer (16 June 2015) and in winter (13 January 2016) in the plant association *Festucetum vaginatae* Rapaics ex Soó 1929 em. Borhidi 1996 (Borhidi *et al.* 2012) lowland steppe open sandy perennial grassland (Pannonian psammophytic grasslands) area Tece of Vácrátót. Voucher specimens (VBI-L 6104, 6105) are deposited in the VBI Lichen Herbarium (Vácrátót, Hungary). The thalli were dried, cleaned and randomised at room temperature (24–26 °C), then 2–4 entire thalli or parts of thalli (20–25

mm × 20–40 mm × 8–12 mm) were placed in 50 ml Falcon tubes where acetone rinsing was applied. Lichens were kept in dry, inactive condition before acetone rinsing. The dry weight of lichen samples (*ca* 1 g) and also the volume of the acetone (40 ml) were measured for each replicates (of 6/treatments used for HPLC analyses – see under HPLC method). The durations of soaking (13 different treatments) were the same as applied by Solhaug and Gauslaa (2001), so that is between 30 min and nearly 6 weeks: 0.5, 1, 2, 4, 8, 16, 32, 64, 128, 256, 384, 640 and 1,024 hrs. After removing the thalli from acetone, the soaking acetone-solution was analysed by HPLC.

HPTLC method

Collected samples were checked for presence of usnic acid and fumarprotocetraric acid by high performance thin layer chromatography (HPTLC) according to standard methods for analysing lichen samples described by Arup *et al.* (1993) and Molnár and Farkas (2011). CAMAG horizontal chamber of 10 cm × 10 cm, CAMAG TLC Plate Heater III, 10 cm × 10 cm thin-layer chromatographic plates (Merck, Kieselgel 60 F254) were used. Solvent system C (toluene–acetic acid, 20:3) was applied.

HPLC method

LSM containing acetone solutions remained after acetone rinsing treatment were analysed by high performance liquid chromatography with photodiode array detector HPLC-PDA. We used a Waters HPLC (e2695) equipped with Photodiode Array Detector (2998 PDA Detector).

Six replicates were applied for each of the 13 treatments (different duration of time for soaking in acetone). The acetone-rinsed lichen thalli were used for fluorescence measurements and LSMs originating from the same thalli were analysed from the soaking acetone. In this way acetone-solution of randomised entire thalli and parts of thalli were sampled. According to our aim we intended to check LSM content of the same thalli – which were used for chlorophyll fluorescence measurement – by the analysis of the soaking acetone-solution.

We tested acetone extraction also on pulverised homogenised samples according to protocol of preparative pharmaceutical analytical investigations. Acetone is a strong organic solvent and it was not surprising that there was no difference between samples extracted for 15, 30, 40 and 50 min. Homogenisation was carried out on lyophilised samples – for 72 hr lyophilisation a freeze dryer Christ Beta 1-8 (with vacuum pump Edwards RV5) was used adjusting condenser temperature to –50 °C, vacuum to 3×10^{-2} mbar.

The following chromatographic conditions were applied for measuring usnic acid and fumarprotocetraric acid (modified design after Ji and Khan 2005).

Chromatographic separation was carried out on a Phenomenex Luna, 4.6×150 mm, $5 \mu\text{m}$ column. The column oven temperature was 40°C and the sample cooler temperature was 5°C . For elution, 0.5% (v/v) ortho-phosphoric acid in water (mobile phase A) and 0.5% (v/v) ortho-phosphoric acid in acetonitrile (mobile phase B) were used as solvents at a flow rate of 1 mL/min.

The duration of the gradient program was 45 min. The gradient program started at 40% B, and after 2 min of isocratic run, solvent B was increased linearly and reached 99.5% at 5 min. Finally, 99.5% B was kept constant for 5 min before getting back to initial conditions to equilibrate the LC before the next injection. Two detection wavelengths $\lambda = 280$ nm for usnic acid and $\lambda = 240$ nm for fumarprotocetraric acid were chosen based on their UV-Vis spectra. Under the described chromatographic conditions the retention times of the compounds of interest were 6.36 min for fumarprotocetraric acid and 15.24

Fig. 1. *Cladonia foliacea* lichen thallus with usnic acid in upper cortex and fumarprotocetraric in lower medullary layer

min for usnic acid. The sample injection volume was 10 μ L. Before injection to the HPLC the samples were filtered (Cronus 25 mm PTFE Syringe Filter 0.22 μ m). For calibration standard stock solutions (1,000 μ g/mL) were prepared from usnic acid and fumarprotocetraric acid reference standards dissolved in acetone. Usnic acid and fumarprotocetraric acid were quantified based on a five-point (5, 10, 20, 50, 100 μ g/mL) calibration.

Statistics

One-way ANOVA followed by Tukey HSD post-hoc tests were applied for LSM concentrations to compare the treatment groups (by seasons). Pearson's rank correlation (R Core Team, 2013) was applied for analysing concentration values of LSMs.

RESULTS AND DISCUSSION

Two major LSMs, depsidone fumarprotocetraric acid and dibenzofuran usnic acid are produced by *C. foliacea* (Hillmann and Grummann 1957, Huneck and Yoshimura 1996, Smith *et al.* 2009, Wirth *et al.* 2013). The presence of these substances was checked by high-performance thin-layer chromatography (HPTLC). Quantitative analysis was carried out by high-performance liquid chromatography with photodiode array detector (HPLC-PDA).

The concentration of usnic acid measured in a series of acetone rinsed thalli was somewhat higher in winter than in summer collected samples (Fig. 2). The concentrations varied between 7.34 and 15.52 mg/g in summer collected samples and 13.90 and 21.61 mg/g in winter collected samples. Variance analysis ($p = 0.05$) showed significant difference in winter samples only at a few (3 out of 13) treatments, in summer samples at somewhat more (7 out of 13) treatments (Table 1). In majority the concentrations in the samples did not differ significantly within treatments. However, significant difference was found between the summer and winter samples in majority (11 out of 13) of the treatments (Table 3).

A comparable amount (11.61 ± 0.29 mg/g) was measured also in the lyophilised, homogenised samples collected in summer.

The higher amount of UV protecting usnic acid measured in winter collected samples might be explained by the longer humid periods accompanied by higher amount of incoming light in this period of year compared to summer environmental conditions (cf. Ertl 1951). The photobionts need more intense protection against harmful light in hydrated, unfolded thallus, therefore the production of this type of LSM might be increased (cf. Gauslaa *et al.* 2017, Vráblíková *et al.* 2006). In contrast lichens are photosynthetically active only

Table 1
P-values of Tukey HSD post-hoc tests for comparison of usnic acid concentration results in summer collected samples (upper right values) and in winter collected samples (lower left values) of *C. foliacea* within 13 treatments (0.5, 1, 2, 4, 8, 16, 32, 64, 128, 256, 384, 640 and 1,024 hrs long acetone rinsing). Statistically significant differences ($p < 0.05$) are marked by boldface

Length of treatment (hrs)	0.5	1	2	4	8	16	32	64	128	256	384	640	1,024	summer
0.5		1.00	0.75	0.69	0.07	0.01	0.06	0.16	0.05	0.26	1.00	0.11	0.48	
1	1.00		0.34	0.29	0.02	0.00	0.01	0.04	0.01	0.07	0.97	0.03	0.16	
2	1.00	0.98		1.00	0.94	0.41	0.91	0.99	0.89	1.00	0.98	0.98	1.00	
4	0.10	0.04	0.44		0.96	0.47	0.94	1.00	0.92	1.00	0.97	0.99	1.00	
8	1.00	1.00	1.00	0.24		1.00	1.00	1.00	1.00	1.00	0.27	1.00	1.00	
16	1.00	1.00	0.99	0.06	1.00		1.00	0.97	1.00	0.90	0.04	0.99	0.68	
32	0.98	0.88	1.00	0.69	1.00	0.94		1.00	1.00	1.00	0.22	1.00	0.99	
64	0.57	0.32	0.97	1.00	0.85	0.42	1.00		1.00	1.00	0.47	1.00	1.00	
128	0.24	0.21	0.91	1.00	0.73	0.29	0.99	1.00		1.00	0.20	1.00	0.99	
256	0.25	0.11	0.76	1.00	0.51	0.16	0.93	1.00	1.00		0.64	1.00	1.00	
384	0.88	0.65	1.00	0.90	0.99	0.76	1.00	1.00	1.00	0.99		0.38	0.87	
640	0.03	0.01	0.20	1.00	0.09	0.02	0.39	0.92	0.97	1.00	0.65		1.00	
1,024	0.05	0.02	0.29	1.00	0.15	0.03	0.51	0.97	0.99	1.00	0.77	1.00		
winter														

Table 2

P-values of Tukey HSD post-hoc tests for comparison of fumarprotocetraric acid concentration results in summer collected samples (upper right values) and in winter collected samples (lower left values) of *C. foliacea* within 13 treatments (0.5, 1, 2, 4, 8, 16, 32, 64, 128, 256, 384, 640 and 1,024 hrs long acetone rinsing). Statistically significant differences ($p < 0.05$) are marked by boldface

Length of treat- ment (hrs)	0.5	1	2	4	8	16	32	64	128	256	384	640	1,024	sum- mer
0.5		1.00	1.00	1.00	1.00	1.00	1.00	1.00	1.00	1.00	1.00	1.00	1.00	0.26
1	0.94		1.00	1.00	1.00	1.00	1.00	1.00	1.00	1.00	1.00	1.00	1.00	0.29
2	0.95	1.00		1.00	1.00	1.00	1.00	1.00	1.00	1.00	1.00	1.00	1.00	0.25
4	1.00	0.91	0.93		1.00	1.00	1.00	1.00	1.00	1.00	1.00	1.00	1.00	0.26
8	0.69	1.00	1.00	0.63		1.00	1.00	1.00	1.00	1.00	1.00	1.00	1.00	0.28
16	0.36	1.00	0.99	0.31	1.00		1.00	1.00	1.00	1.00	1.00	1.00	1.00	0.34
32	0.74	1.00	1.00	0.68	1.00	1.00		1.00	1.00	1.00	1.00	1.00	1.00	0.28
64	0.44	1.00	1.00	0.39	1.00	1.00	1.00		1.00	1.00	1.00	1.00	1.00	0.31
128	0.85	1.00	1.00	0.81	1.00	1.00	1.00	1.00		1.00	1.00	1.00	1.00	0.32
256	0.99	1.00	1.00	0.99	1.00	0.95	1.00	0.98	1.00		1.00	1.00	1.00	0.35
384	0.16	0.93	0.91	0.13	1.00	1.00	0.99	1.00	0.98	0.76		1.00	1.00	0.28
640	1.00	0.99	0.99	1.00	0.87	0.56	0.90	0.66	0.96	1.00	0.29	0.31		
1,024	1.00	1.00	1.00	1.00	1.00	0.91	1.00	0.95	1.00	1.00	0.67	1.00		
winter														

Table 3

P-values testing differences between the usnic acid (mg/g) and fumarprotocetraric acid content (mg/g) of summer and in winter collected *C. foliacea* thalli after 0.5, 1, 2, 4, 8, 16, 32, 64, 128, 256, 384, 640 and 1,024 hrs long acetone rinsing. Statistically significant differences are marked with asterisks (***) = $p < 0.05$, ** = $0.1 < p < 0.05$); n = 6

Length of treatment (hrs)	P _{usnic acid}	P _{fumarprotocetraric acid}	
0.5	0.896	0.103	
1	0.365	0.003	***
2	0.037	0.001	***
4	0.000	0.043	***
8	0.026	0.043	***
16	0.011	0.006	***
32	0.005	0.053	**
64	0.005	0.034	***
128	0.001	0.157	
256	0.005	0.674	
384	0.023	0.009	***
640	0.001	0.749	
1,024	0.037	0.810

Fig. 2. Relation between usnic acid content (mg/g) and the duration of acetone rinsing (hrs) of *Cladonia foliacea* samples collected in summer (n = 6) and winter (n = 6) (error bars show standard deviations)

for a few hours early in the morning in summer according to former measurements carried out also in Hungarian Great Plain (Kalapos and Mázsa 2001, Lange *et al.* 1970, Mázsa 1994, Mázsa *et al.* 1998, 1999). At this time thalli are wet because of dawn dew, but a few hours after sunrise they dry out quickly (Kalapos and Mázsa 2001, Lange *et al.* 1970, Verseghe *et al.* 1987). Thalli needed less intense photoprotection, and fewer amounts of UV protecting metabolites against harmful excess light. Bjerke *et al.* (2005), finding 51.1 to 53 mg/g usnic acid in Arctic, alpine and Patagonian populations of another species (*Flavocetraria nivalis* (L.) Kärnefelt et A. Thell), reported large variations among samples. BeGora and Fahselt (2001) found that usnic acid production in *Cladonia arbuscula* subsp. *mitis* (Sandst.) Burgaz was the lowest at sampling times when the highest UV-B radiation was detected.

There is a similar tendency in fumarprotocetraric acid concentrations (Fig. 3): the winter collected values are above the summer collected values at each treatment. The concentrations varied between 0.60 and 3.01 mg/g in summer collected samples and 2.26 and 5.81 mg/g in winter collected samples. Neither summer, nor winter values differed significantly within the 13 treatments (Table 2). However, significant difference was found between the summer and winter treatments in majority (7 out of 13) of the treatments (Table 3).

A comparable amount (2.45 ± 0.21 mg/g) was measured also in the lyophilised, homogenised samples collected in summer.

Fig. 3. Relation between fumarprotocetraric acid content (mg/g) and the duration of acetone rinsing (hrs) of *Cladonia foliacea* samples collected in summer ($n = 6$) and winter ($n = 6$) related to (error bars show standard deviations)

These values are very similar to data (0.11–5.4 mg/g) given by Archer (1981) for Australian scyphose *Cladonia* species (collected between October 1978 and February 1980).

There is a significant difference between concentrations of the summer and winter collected samples in respect of both usnic and fumarprotocetraric acids (Table 3). In case of usnic acid concentration values differed considerably, except for short treatments (0.5–1 hr). Collecting season is somewhat more important for fumarprotocetraric acid, if short treatment is applied, however at the long (128–1,024 hrs) treatments with acetone, fumarprotocetraric acid contents did not differ significantly.

In the same time there is a weak negative correlation between the usnic acid and fumarprotocetraric acid concentration values both in summer ($r^2 = 0.28$) and in winter ($r^2 = 0.17$). This result might be caused by the special morphology of *C. foliacea* thalli changing with humidity conditions. If samples (placed in a single Falcon tube) contained more thalli, parts of thalli which was effected by irradiance on their upper surface (usnic acid containing upper cortex) for longer time, the concentration of usnic acid was relatively higher and that of fumarprotocetraric acid was lower. If there was relatively more of the enrolled pale lower surface (where the production of fumarprotocetraric acid is characteristic in the medulla) which was effected by irradiance for a longer time, a slightly higher fumarprotocetraric acid content could be detected compared to the amount of usnic acid. The moderate variability within treatments found by variance analysis justifies that HPLC measurements of the soaking acetone provides valuable results.

The detrimental effect of acetone tested by chlorophyll fluorescence measurement resulted in tolerance until *ca* 5 days (Farkas *et al.* 2020). The slight seasonal difference between the summer and winter collected thalli is not significant in this respect.

HPLC measurements show moderate variability for usnic acid and no significant variability for fumarprotocetraric acid in the measured values, thus variance analysis ($p = 0.05$) resulted in significant difference both within the winter and summer samples only at a few treatments in usnic acid concentrations. However there is a significant difference between concentrations of summer and winter collected samples in respect of both usnic and fumarprotocetraric acids. Thalli contain relatively more usnic acid and fumarprotocetraric acid in winter than in summer. It must be considered, that while little is known about the metabolism of these substances, it was justified (P.-Verseghe 1976) that the thickness of thalline layers of some temperate xerophilous *Cladonia* species (incl. *C. foliacea* – as *C. convoluta* in the cited paper) is changing by seasons. Therefore it may be supposed that the summer and winter collected thalli were different in the terms of their thalline layer com-

position. Furthermore the moderate water solubility of fumarprotocetraric acid has been observed by us during HPTLC studies, therefore this substance must have a considerable dynamics during the shift of rainy and dry periods (Burgaz, pers. comm.).

The usnic acid containing cortex protects the photobionts from UV irradiation, since usnic acid may act as UV screening agent or have solar radiation protective function (Galloway 1993, McEvoy *et al.* 2007, Nguyen *et al.* 2013, Nybakken *et al.* 2004, 2007). It seems so from the results, that a higher amount of usnic acid is produced in winter to ensure sufficient protection also for summer due to the less amount of harmful incident light. The UV protecting role has not been reported for fumarprotocetraric acid. However, according to our hypothesis the fumarprotocetraric acid content of the medulla might contribute to the solar irradiation reflecting role of the pale lower surface lobes turning upwards in dry condition. Since the range of concentration values originating from the analysis of the soaking acetone-solution is comparable with amounts found in pulverised samples, as well as with data known from lichens previously, these data represent valuable contribution to the knowledge on production of these LSMs in this particular species and forms a basis for further comparative investigations.

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